

Problem: 9-79

Given: An ideal air-standard diesel cycle with a compression ratio of 15,
inlet conditions: $70^\circ F$ and 1 atm,
700 Btu/lbm heat input

Find: Pressure and temperature after each process and cycle efficiency

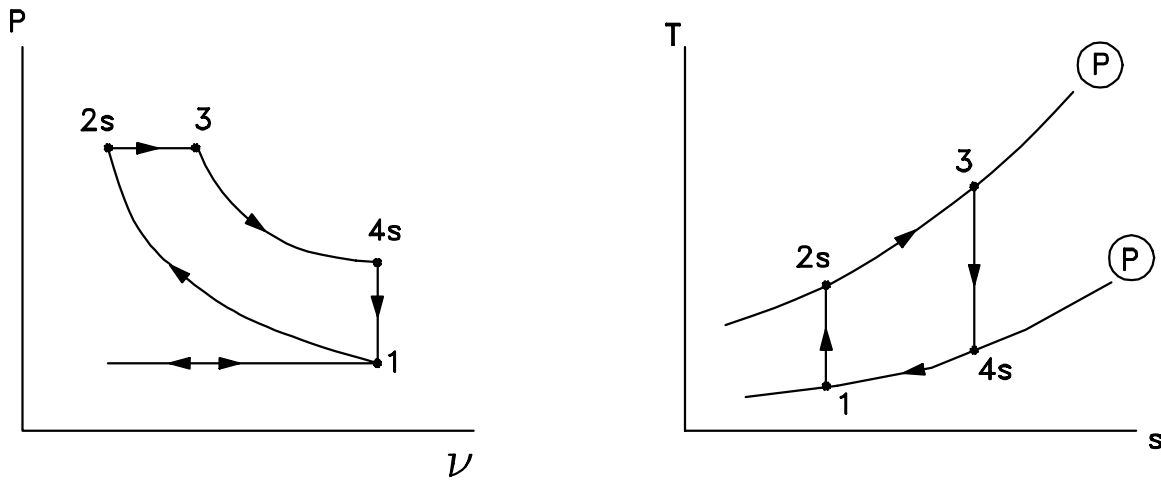


Table of values

	T (R)	P (psi)	ν (ft^3/lbm)
1	530	14.7	13.36
2s	1566	651.4	
3	4482	651.4	2.55
4s	2310	64.1	13.36

Assumptions

- ideal gas, $c_p = \text{constant}$
- reversible
- $\Delta ke = \Delta pe = 0$
- closed system

- adiabatic compression and expansion

1 → 2 isentropic compression

$$P_{2s} = P_1 \left(\frac{\nu_1}{\nu_{2s}} \right)^k = P_1 \cdot r^k = 14.7(15)^{1.4} = 651.4 \text{ psi}$$

$$T_{2s} = T_1 \left(\frac{\nu_1}{\nu_{2s}} \right)^{k-1} = T_1 \cdot r^k = 530(15)^{.4} = 1566 \text{ R}$$

2 → 3 constant pressure (heat added)

$$P_3 = P_{2s} = 651.4 \text{ psi}$$

$$Q_{in} = h_3 - h_{2s} = c_p(T_3 - T_{2s}) \Rightarrow T_3 = \frac{Q_{in}}{c_p} + T_{2s} = \frac{700}{.24} + 1566 = 4482 \text{ R}$$

$$\nu_3 = RT_3/P_3 = \frac{(53.34 \text{ ft} \cdot \text{lb f}/(\text{lb m} \cdot \text{R}))4482 \text{ R}}{651.4 \text{ psi}} \times \frac{\text{ft}^3}{144 \text{ in}^2} = 2.55 \text{ ft}^3/\text{lb m}$$

3 → 4 isentropic expansion

$$P_{4s} = P_3 \left(\frac{\nu_3}{\nu_{4s}} \right)^k$$

$$= (651.4)(2.55/13.36)^{1.4} = 64.1 \text{ psi}$$

where

$$\nu_{4s} = \nu_1 = \frac{RT_1}{P_1} = (53.34) \frac{(530)}{14.7} \cdot \frac{1}{144} = 13.36 \text{ ft}^3/\text{lb m}$$

$$T_{4s} = T_3 \left(\frac{\nu_3}{\nu_{4s}} \right)^{k-1} = (4482) \left(\frac{2.55}{13.36} \right)^{0.4} = 2310 \text{ R}$$

$$\eta = \frac{\dot{W}_{net}}{\dot{Q}_{in}} = 1 - \frac{Q_{out}}{Q_{in}}$$

where

$$Q_{out} = (u_4 - u_1) = c_v(T_4 - T_1) = 0.172(2310 - 530)$$

$$= 306.2 \text{ Btu/lb m}$$

$$Q_{in} = 700 \text{ Btu/lb m}$$

$$\eta = 1 - \frac{306.2}{700} = \boxed{56.3\%}$$